

(3E)-3-[(4-Butylphenyl)imino]-1,3-dihydro-2H-indol-2-one

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Key indicators

Single-crystal X-ray study
 $T = 294\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$
 R factor = 0.049
 wR factor = 0.128
Data-to-parameter ratio = 15.4

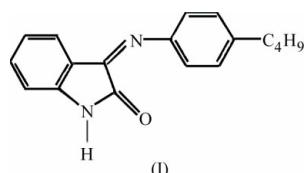
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, $C_{18}H_{18}N_2O$, has a non-planar conformation. The indol and butylphenyl groups are connected by a C—N bond [1.433 (3) Å]. The crystal structure is stabilized by intermolecular N—H···N and C—H···O interactions.

Received 22 April 2003
Accepted 30 April 2003
Online 9 May 2003

Comment

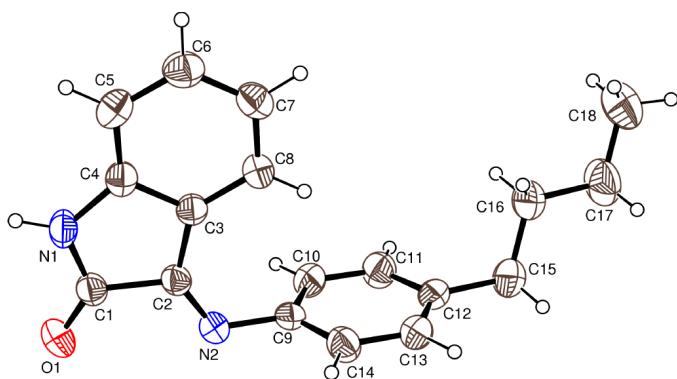
Isatin and its derivatives have been used as reagents in the dye industry. Schiff bases of isatin were reported to possess anti-HIV (Pandeya *et al.*, 2000), antifungal (Pandeya *et al.*, 1999), antibacterial (Sarangapani & Reddy, 1994; Varma & Nobles, 1975), antiviral (Singh *et al.*, 1983), antiprotozoal (Varma & Khan, 1977) and antihelminthic (Sarciron *et al.*, 1993) activities. The medical and biological implications of this category of ligands has already been proved (Popp & Pajouhesh, 1982).



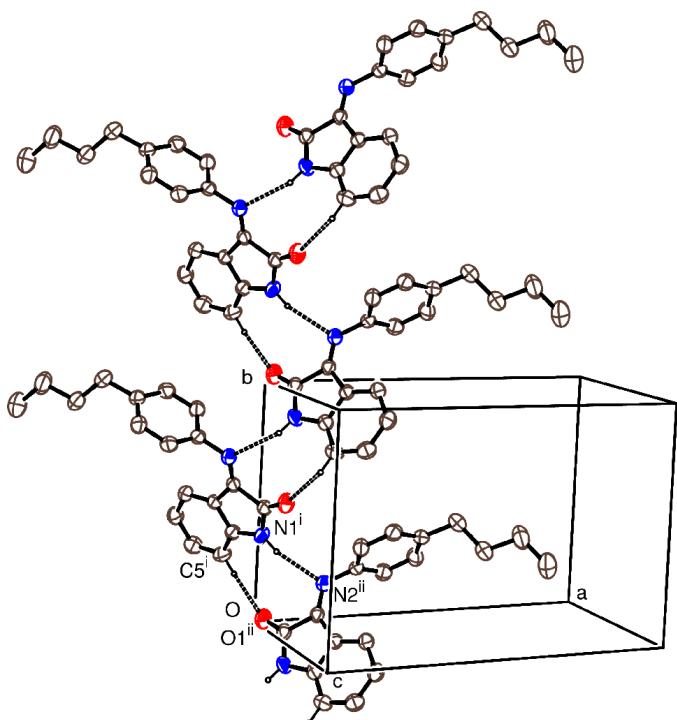
The structure of the title compound, (I), is shown in Fig. 1. The C1—C2 bond length [1.529 (3) Å] is within the range 1.49–1.56 Å observed for related compounds found in the Cambridge Structural Database (Allen, 2002). The C2—N2—C9 angle is 119.6 (2)°. In the butyl group, the average C—C—C bond angle is 114.7 (3)° and this group shows an *E* form. The indole group is planar [maximum displacement is 0.004 (2) Å for C1] and forms a dihedral angle of 89.8 (1)° with the phenyl plane. These bond distances and angles agree with the values reported for (3*E*)-3-[(4-hexylphenyl)imino]-1*H*-indol-2(3*H*)-one (Özürk *et al.*, 2003).

The N—H···N and C—H···O hydrogen bonds form zigzag chains, parallel to the *b* axis (Fig. 2). The geometry of the hydrogen bonds is given in Table 2.

To determine the structural and electronic parameters of (I), quantum-chemical calculations were carried out using the PM3 method (Stewart, 1985). It was found that the charges at atoms O1, N1 and N2 are 0.0382, 0.0609 and -0.2930 e^- , respectively. The final heat of formation of (I) is 14.98 kcal and its total energy is -3027.82 eV . The energies of the HOMO and LUMO levels have the values -9.0903 and -0.9315 eV , respectively. The calculated molecule dipole moment is 4.352 Debye.

**Figure 1**

The molecular structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme.

**Figure 2**

A view of the intermolecular hydrogen-bond contacts, showing the zigzag chain which develops parallel to b . [Symmetry codes: (i) $-x, -y, -z$; (ii) $x, \frac{1}{2} - y, \frac{1}{2} + z$.]

Experimental

The title compound was prepared according to the method of Öztürk *et al.* (2003). The orange product was recrystallized from methanol (m.p. 451–458 K).

Crystal data

$C_{18}H_{18}N_2O$
 $M_r = 278.34$
Monoclinic, $P2_1/c$
 $a = 15.6069 (2)$ Å
 $b = 9.5596 (2)$ Å
 $c = 10.5265 (2)$ Å
 $\beta = 107.187 (2)^\circ$
 $V = 1500.38 (5)$ Å 3
 $Z = 4$

$D_x = 1.232$ Mg m $^{-3}$
Mo $K\alpha$ radiation
Cell parameters from 171 reflections
 $\theta = 6.0$ – 26.0°
 $\mu = 0.08$ mm $^{-1}$
 $T = 294 (2)$ K
Slab, orange
0.40 × 0.31 × 0.17 mm

Data collection

Nonius KappaCCD diffractometer
 ω scans
Absorption correction: none
9152 measured reflections
2938 independent reflections
1851 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.072$
 $\theta_{\text{max}} = 26.0^\circ$
 $h = -19 \rightarrow 19$
 $k = -11 \rightarrow 11$
 $l = -12 \rightarrow 12$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.049$
 $wR(F^2) = 0.128$
 $S = 1.02$
2938 reflections
191 parameters
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0558P)^2$
+ 0.2681 $P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.17$ e Å $^{-3}$
 $\Delta\rho_{\text{min}} = -0.21$ e Å $^{-3}$

Table 1
Selected geometric parameters (Å, °).

O1—C1	1.214 (2)	C1—C2	1.531 (3)
N1—C1	1.356 (3)	C15—C16	1.524 (3)
N1—C4	1.413 (2)	C16—C17	1.502 (3)
N2—C2	1.272 (2)	C17—C18	1.504 (4)
N2—C9	1.433 (2)		
C1—N1—C4	111.73 (16)	C5—C4—N1	128.15 (18)
C2—N2—C9	119.60 (16)	C3—C4—N1	109.73 (17)
O1—C1—N1	128.03 (18)	C10—C9—N2	118.27 (18)
O1—C1—C2	125.99 (19)	C14—C9—N2	121.99 (19)
N1—C1—C2	105.91 (16)	C17—C16—C15	114.9 (2)
N2—C2—C3	135.31 (17)	C16—C17—C18	114.5 (2)
N2—C2—C1	118.87 (17)		

Table 2
Hydrogen-bonding geometry (Å, °).

$D—H \cdots A$	$D—H$	$H \cdots A$	$D \cdots A$	$D—H \cdots A$
N1—H1 \cdots O1 ⁱⁱⁱ	0.86	2.24	3.062 (2)	159
C5—H5 \cdots O1 ⁱⁱⁱ	0.93	2.56	3.254 (3)	132

Symmetry code: (iii) $-x, \frac{1}{2} + y, \frac{3}{2} - z$.

The H atoms of C—H and N—H groups were placed in calculated positions (C—H = 0.96 Å and N—H = 0.86 Å) and were allowed to refine as riding models, with U_{iso} set equal to $1.2U_{\text{eq}}$ (1.5 for CH₃) of the carrier atoms.

Data collection: *COLLECT* (Nonius, 1999); cell refinement: *EVALCCD* (Duisenberg, 1998); data reduction: *EVALCCD*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997; Burnett & Johnson, 1996); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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